

# **GATE – Mining Engineering**

**(Topic Wise Questions 2007-2017)**

**Topic: Surface Mine**

**Environment**

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## **GATE SYLLABUS:**

### **Ventilation, Underground Hazards and Surface Environment:**

Underground atmosphere; Heat load sources and thermal environment, air cooling; Mechanics of air flow, distribution, natural and mechanical ventilation; Mine fans and their usage; Auxiliary ventilation; Ventilation planning.

Subsurface hazards from fires, explosions, gases, dust and inundation; Rescue apparatus and practices; Safety in mines, accident analysis, noise, mine lighting, occupational health and risk.

Air, water and soil pollution: causes, dispersion, quality standards, reclamation and control.

## 2007

- Q.11 In the Gaussian plume model, the dispersion coefficients are function of
- (A) distance from source and stability class
  - (B) stack height and distance from source
  - (C) stability class and source coordinates
  - (D) source coordinates and distance from source
- Q.16 The most recent model of self-contained compressed-oxygen breathing apparatus is
- (A) Proto-IV                      (B) BG-174                      (C) BG-4                      (D) BG-174A
- Q.52 An effluent sample is diluted with fresh water to make up a solution of 300 ml. The DO of the solution initially is 8.0 mg/l and the value falls to 3.0 mg/l after 5 days. If the 5-day BOD of the original effluent is known to be 50 mg/l, the amount of fresh water added in ml to the solution is
- (A) 270                      (B) 160                      (C) 54                      (D) 30
- Q.53 With respect to stack emission the phenomenon of fumigation is noticed in case of
- (A) atmospheric lapse rate being lower than the adiabatic lapse rate
  - (B) atmospheric lapse rate being higher than the adiabatic lapse rate
  - (C) temperature inversion in the atmosphere above the stack height
  - (D) temperature inversion in the atmosphere below the stack height

## 2008

- Q.16 Electrostatic precipitator works on the principle of
- (A) Capacitance change
  - (B) Ionization of the particles
  - (C) Electro heating of gases
  - (D) Centrifuging the gaseous molecules
- Q.35 Precipitation of metallic ions in mine water drainage is carried out by
- (A)  $\text{CaSO}_4$  and  $\text{MgSO}_4$
  - (B)  $\text{CaCO}_3$  and  $\text{MgCO}_3$
  - (C)  $\text{Ca(OH)}_2$  and  $\text{NaOH}$
  - (D)  $\text{CaCO}_3$  and  $\text{MgSO}_4$

Q.44 A roadheader district produces  $20 \text{ mg/m}^3$  of airborne dust with the following size distribution:

Size up to	Cumulative wt %
1 $\mu\text{m}$	1
5 $\mu\text{m}$	5
10 $\mu\text{m}$	10
20 $\mu\text{m}$	20
50 $\mu\text{m}$	50
> 50 $\mu\text{m}$	100

The concentration of respirable fraction of dust in  $\text{mg/m}^3$  is

- (A) 0.2                      (B) 2.0                      (C) 10.0                      (D) 1.0

**Statement for Linked Answer Questions 78 and 79:** Mine water flowing at  $1.5 \text{ m}^3/\text{s}$  with  $2 \text{ mg/l}$  dissolved oxygen, joins river water flowing at  $7 \text{ m}^3/\text{s}$  containing  $6 \text{ mg/l}$  dissolved oxygen.

Q.78 The dissolved oxygen concentration of the mixture in  $\text{mg/l}$  is

- (A) 5.3                      (B) 4.8                      (C) 4.2                      (D) 3.9

Q.79 The saturated value of the dissolved oxygen in the mixture is given to be  $9.3 \text{ mg/l}$ . On this basis, the initial oxygen deficit of the mixture in  $\text{mg/l}$  is

- (A) 2.4                      (B) 4.0                      (C) 6.8                      (D) 14.6

## 2009

Q.15 Moody diagram represents resistance coefficient in terms of

- (A) Reynolds number and asperity ratio                      (B) viscosity and aspect ratio  
(C) surface tension and viscosity                      (D) Reynolds number and surface tension

Q.48 In an area within a surface mine, under static condition the following gases are found:  $\text{NO}_2$ ,  $\text{CO}_2$ ,  $\text{O}_3$  and  $\text{SO}_2$ . Assuming no diffusion, reaction and bonding of the gases, the concentration of the gases from bottom upwards will be in the order of

- (A)  $\text{NO}_2$ ,  $\text{CO}_2$ ,  $\text{O}_3$  and  $\text{SO}_2$   
(B)  $\text{SO}_2$ ,  $\text{NO}_2$ ,  $\text{CO}_2$  and  $\text{O}_3$   
(C)  $\text{SO}_2$ ,  $\text{O}_3$ ,  $\text{NO}_2$  and  $\text{CO}_2$   
(D)  $\text{NO}_2$ ,  $\text{CO}_2$ ,  $\text{SO}_2$  and  $\text{O}_3$

## 2011

- Q.11 The whole circle bearing of the line AB is  $116^{\circ}20'20''$ . If there exists an east declination of  $20^{\circ}$ , the true quadrantal bearing of line AB is  
 (A)  $S41^{\circ}59'40''E$  (B)  $S43^{\circ}39'40''E$  (C)  $S45^{\circ}59'40''W$  (D)  $S47^{\circ}59'40''W$
- Q.12 It is proposed to connect two straights of a road by a simple circular curve. If the maximum speed of the vehicle is 60 km/h and the centrifugal ratio for the road is  $1/4$ , the minimum radius of the curve in m is  
 (A) 113.26 (B) 98.18 (C) 25.46 (D) 15.50
- Q.20 ISO 9000 Quality Systems **DO NOT** contain  
 (A) legal provisions (B) measurement (C) document control (D) standardization
- Q.23 100 ml of waste water is allowed to evaporate in a dish weighing 48.6232 g. The weight of the dish with dry solids is 48.6432 g. The concentration of dry solids in waste water in mg/l is  
 (A) 200 (B) 220 (C) 260 (D) 320

## 2012

- Q.35 A spherical droplet of water, with density  $1000 \text{ kg/m}^3$  and diameter of  $1 \mu\text{m}$ , is falling in air. The viscosity of air is  $1.85 \times 10^{-5} \text{ kg/m}\cdot\text{s}$ . Neglecting air density and assuming that the settling of droplet in air follows Stokes' Law, the settling velocity in m/s is  
 (A)  $0.98 \times 10^{-5}$  (B)  $2.95 \times 10^{-5}$  (C)  $8.04 \times 10^{-5}$  (D)  $53.03 \times 10^{-5}$

## 2013

- Q.27 In an experiment to determine specific gravity of a soil sample, the following data is obtained:
- |   |        |
|---|--------|
| Mass of empty pycnometer                              | 20.4 g |
| Mass of pycnometer with soil sample                   | 51.6 g |
| Mass of pycnometer with soil sample filled with water | 88.6 g |
| Mass of pycnometer filled with water                  | 70.4 g |

The specific gravity of the sample is \_\_\_\_\_

## 2014

- Q.20 A mine waste dump of pH 5.2 can be neutralized by adding  
 (A) urea (B) calcium carbonate  
 (C) sulphuric acid (D) sodium chloride

## 2015

Question Number : 16 Question Type : MCQ

The temperature of a parcel of air decreases from  $30.2^{\circ}\text{C}$  to  $28.9^{\circ}\text{C}$  as it rises from an altitude of 20 m to 120 m. The lapse rate for the atmosphere is

- (A) subadiabatic      (B) adiabatic      (C) superadiabatic      (D) transadiabatic

Options :

1. ✘ A
2. ✘ B
3. ✔ C
4. ✘ D

Question Number : 23 Question Type : MCQ

In a BOD test, 5 ml of wastewater is diluted with pure water to fill a 300 ml BOD bottle. The initial and final dissolved oxygen contents of the mix are 9.0 mg/l and 7.0 mg/l respectively. The BOD of the wastewater, in mg/l, is

- (A) 2      (B) 10      (C) 120      (D) 600

Options :

1. ✘ A
2. ✘ B
3. ✔ C
4. ✘ D

## 2016

Q.21 Identify the WRONG statement.

The 'temperature inversion' of the atmosphere in surface mines aggravates the problem of

- (A) airborne dust  
(B) noise  
(C) ground vibrations  
(D) visibility

# 2017

**Question Number : 22**

**Correct : 1 Wrong : -0.33**

The “yellow boy” formed due to acid mine drainage mainly consists of

- (A) Ferrous hydroxide
- (B) Ferrous sulfate
- (C) Ferric hydroxide
- (D) Ferric sulfate

**Question Number : 50**

**Correct : 2 Wrong : 0**

A stream flowing at  $15 \text{ m}^3/\text{s}$  has a tributary feeding into it with a flow rate of  $7 \text{ m}^3/\text{s}$ . The concentrations of chloride at the upstream of the junction and that of the tributary are  $30 \text{ mg/L}$ , and  $50 \text{ mg/L}$  respectively. Treating chloride as conservative substance and assuming complete mixing of two streams, the concentration of chloride in  $\text{mg/L}$  at the downstream is \_\_\_\_\_

**Question Number : 51**

**Correct : 2 Wrong : 0**

A sample of mine water has  $100 \text{ mg/L}$  of  $\text{Ca}^{2+}$  and  $10 \text{ mg/L}$  of  $\text{Mg}^{2+}$ . The equivalent weights of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  are  $20 \text{ mg/meq}$  and  $12.2 \text{ mg/meq}$  respectively. The hardness of mine water in unit of  $\text{mg/L}$  as  $\text{CaCO}_3$  is \_\_\_\_\_