

An Updated and Extended Mohs Mineral Hardness Scale

Friedrich Mohs (1773-1839), an Austrian/German mineralogist, developed a scale of hardness for minerals in 1812 (although some sources reference 1822). This hardness scale classifies minerals on a scale from one (the softest) to ten (the hardest) and has been used by mineralogists since its inception. The minerals Friedrich Mohs selected for his scale were natural, pure substances which were commonly or readily available at the time. Basically – it tests to see how easily one substance can scratch another. For example; if your material to be tested can be scratched by Fluorite (hardness of 4) but not scratched by Calcite (hardness of three), then your test material would have a hardness of 3.5 on the Mohs scale.

Although the Mohs Mineral Hardness scale appears to be Linear (one to ten), it is not – it's a Relative scale. Which is why the “absolute scale” was developed. If you look at the Mohs scale, it would appear that Corundum (9) is one-half as soft as Diamond (10), when in actuality, Corundum is four-times as soft. That is why scientists who required a more accurate measurement of hardness developed a “proportional measurement” known as the “absolute mineral hardness” to account for this (non-linear) discrepancy in the Mohs traditional scale by using a very sensitive piece of equipment called a “sclerometer. The sclerometer accounts for the large relative difference in hardness as one approaches the higher end of the Mohs scale.

It should be noted that Friedrich Mohs was not the first person to use this method of comparing hardness by observing which mineral scratches another. Indeed – some references suggest that Friedrich Mohs obtained the idea from the miners of his time. Plus, comparing mineral hardness has some roots in antiquity. It was first mentioned by Theophrastus in his treatise *On Stone*, circa 300 BC and later followed by Pliny the Elder in his *Naturalis Historica*, circa 77 AD.

Additionally, the traditional Mohs scale did not account for half-number hardness values. For example; Dolomite will scratch Calcite (hardness of 3) but not Fluorite (hardness of 4). Therefore, how would you classify Dolomite on the traditional Mohs scale since it is neither a 3 or a 4? The answer was to add the half-number values to the scale. Now, Dolomite would be classified as a 3.5 on the hardness scale.

Lastly, the hardness of some minerals vary based upon how the hardness test is performed – with the grain or against the grain. A good example of this is Kyanite, which usually is a long bladed crystal. When you scratch Kyanite across the grain (the short side of the blade) it has a hardness of 7. However, when you scratch it with the grain (longways down the blade) it has a hardness of 4. Muscovite is another good example of this dual-hardness property. Muscovite has a hardness of 2.5 when taken across the flat surface of a cleavage sheet, but has a hardness of 4 when taken across the grain. Basically this dual-hardness phenomenon depends upon the strength of the molecular bond that holds the mineral together.

NOTES:

1. Data taken from Discover Magazine, July/August 2009, Page 12, Physics Beat by Adam Hadhazy. The content of that article is contained within the box that follows...

HOW TO SCRATCH A DIAMOND

The reputation of diamond as the hardest material around is under threat. Researchers in China and the United States recently determined that two naturally occurring substances surpass diamond's resistance to scratching and indentation. They calculated that the mineral *Ionsdaleite* – made of carbon, like diamond – is 58 percent harder than its famous cousin. And, *Wurtzite boron nitride* beats diamond's hardness by about 18 percent after being subjected to pressure, which alters its atomic bonds.

Still, in the short term diamond will continue to dominate in practical applications such as saws, drill bits and industrial abrasives, since the newly studied materials are extremely rare. *Ionsdaleite* forms only under the extreme pressure and heat accompanying meteorite impacts, while *Wurtzite boron nitride* is a by-product of intense volcanic eruption. But scientists can create both substances in the lab, says physicist John Janik at the Carnegie Institution for Science. Although producing the conditions required to grow the substances in bulk remains a challenge, Janik and others are working on synthesizing *Wurtzite boron nitride* and *Ionsdaleite* to pave the way for commercial use.

Garry W. Kappel, President – Sedona Gem & Mineral Club, Inc. (2011)

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Mohs Hardness	Traditional Mineral/Substance	Absolute Hardness	Chemical Composition	Other Associated Minerals
0.5	Lithium			Sodium, Potassium
1.0	Talc	1	Magnesium Silicate Hydroxide Mg₃Si₄O₁₀ (OH)₂	Steatite, Graphine, Soapstone
1.5	Graphite			Gallium, Tin, Barium, Lead
2.0	Gypsum	3	Calcium Sulfate Hydrate CaSO₄ 2H₂O	Amber, Calcium, Sulfur, Bismuth, Stibnite
2.5	Fingernail			Magnesium, Gold, Silver, Zinc, Aluminum, Galena
3.0	Calcite	9	Calcium Carbonate CaCO₃	Copper, Pearl, Antimony
3.5	Copper Penny			Arsenic, Howlite, Coral
4.0	Fluorite	21	Calcium Fluoride CaF₂	Malachite, Iron, Nickle
4.5	Platinum			Steel, Variscite
5.0	Apatite	48	Calcium Phosphate Ca₅(PO₄)₃(OH⁻, Cl⁻, F⁻)	Dioptase, Cobalt, Zirconium, Tooth enamel
5.5	Glass, Knife Blade			Beryllium, Cobalite, Opal, Molybdenum
6.0	Orthoclase Feldspar	72	Potassium Aluminum Silicate KAlSi₃O₈	Hematite, Titanium, Manganese, Uranium
6.5	Steel File			Iron Pyrite, Silicon, Iridium, Nephrite Jade
7.0	Quartz	100	Silicon Dioxide SiO₂	Tourmaline, Vanadium, Osmium, Aventurine
7.5	Garnet Sandpaper			Garnet, Emerald, Tungsten, Aquamarine
8.0	Topaz	200	Aluminum Silicate Fluoride Hydroxide Al₂SiO₄(OH⁻, F⁻)₂	Spinal, Cubic Zirconia
8.5	Alexandrite (Beryllium Aluminum Oxide, BeAl₂O₄)			Chrysoberyl,
9.0	Corundum	400	Aluminum Oxide Al₂O₃	Ruby, Sapphire
9.5	Silicon Carbide			Tungsten Carbide, Stishovite
10.0	Diamond	1600	Carbon C	
11.0 ¹	Wurtzite boron nitride	1888		18 percent Harder than Diamod
12.0 ¹	Ionsdaleite	2528		58 percent Harder than Diamond